

**ACADEMIC PLANNER & UNITIZATION OF SYLLABUS****Department of Chemistry****Bankura Christian College****ACADEMIC YEAR: 2023-24 (Semester 2<sup>nd</sup> /4<sup>th</sup> /6<sup>th</sup>)****2nd Semester (GE) Theory (January,2023 to June, 2023)****Name of faculty member: Dr. Bibekananda Mahanti****Subject: Chemistry**

GE T2 – States of Matter &amp; Chemical Kinetics, Chemical Bonding &amp; Molecular

Structure, P-Block Elements

**Unit – 1.**

2. Comparative study of p-block elements

a. Group trends in electronic configuration, modification of pure elements, common oxidation states, inert pair effect, and their important compounds in respect of the following groups of elements:

i. B-Al-Ga-In-Tl

iii. N-P-As-Sb-Bi

iv

**Unit – 2**

N-P-As-Sb-Bi

C-Si-Ge-Sn-Pb

**Unit – 3**

.O-S-Se-Te

**Unit – 4.**

F-Cl-Br-I

MONTH/YEAR	WEEK	PORTIONS
March 2023	2 <sup>nd</sup>	Comparative study of p-block elements: Group trends in electronic configuration, modification of pure elements, common oxidation states: B-Al-Ga-In-Tl
	3 <sup>rd</sup>	Comparative study of p-block elements: Group trends in inert pair effect, and their important compounds in respect of the following groups of elements: i. B-Al-Ga-In-Tl
	4 <sup>th</sup>	Comparative study of p-block elements: Group trends in electronic configuration, modification of pure elements, common oxidation States: C-Si-Ge-Sn-Pb
	5 <sup>th</sup>	Class Test-1/ Revision of Unt-1
April 2023	1 <sup>st</sup>	Comparative study of p-block elements: Group trends in inert pair effect, and their important compounds in respect of the following groups of elements: C-Si-Ge-Sn-Pb
	2 <sup>nd</sup>	Comparative study of p-block elements: Group trends in electronic configuration, modification of pure elements, common oxidation states N-P-As-Sb-Bi
	3 <sup>rd</sup>	.Comparative study of p-block elements: Group trends in inert pair effect, and their important compounds in respect of the following groups of elements: N-P-As-Sb-Bi
	4 <sup>th</sup>	Class Test-2
May 2023	1 <sup>st</sup>	.Comparative study of p-block elements: Group trends in electronic configuration, modification of pure elements, common oxidation states O-S-Se-Te
	2 <sup>nd</sup>	. Comparative study of p-block elements: Group trends in inert pair effect, and their important compounds in respect of the following groups of elements: O-S-Se-Te
	3 <sup>rd</sup>	Comparative study of p-block elements: Group trends in inert pair effect, and their important compounds in respect of the following groups of

		elements: O-S-Se-Te
	4 <sup>th</sup>	Revision of Unt-3
June 2023	1 <sup>st</sup>	Comparative study of p-block elements: Group trends in electronic configuration, modification of pure elements, common oxidation States:F-Cl-Br-I
	2 <sup>nd</sup>	Comparative study of p-block elements: Group trends in inert pair effect, and their important compounds in respect of the following groups of elements: F-Cl-Br-I
	3 <sup>rd</sup>	Comparative study of p-block elements: Group trends in inert pair effect, and their important compounds in respect of the following groups of elements: F-Cl-Br-I